Electronic Structure of an Atom

- 1. What is the nucleon number for $^{40}_{20}$ Ca?
 - **A** 20

C 10

B 40

D 60

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- 2. The atoms of element Q can be represented by ^a_b Q. If Q has 6 protons and 9 neutrons, what are a and b?

| | a | b |
|---|----|----|
| A | 6 | 9 |
| В | 9 | 6 |
| C | 15 | 6 |
| D | 6 | 15 |

- 3. The atom of element X is represented by ${}^{18}_{8}$ X. What can we deduce about the atomic structure of X?
 - **A** X has 8 electrons and 18 protons.
 - **B** X has 18 neutrons and 18 protons.
 - C X has 8 neutrons and 18 protons.
 - **D** X has 8 electrons and 8 protons.

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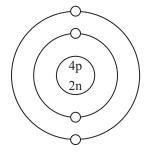
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- 4. What can be deduced from the symbol ${}^{12}_{6}$ C?
 - A An atom of carbon contains 6 electrons in its outermost shell.
 - **B** An atom of carbon has 6 protons and 6 neutrons in its nucleus.
 - C Carbon has a proton (atomic) number of 12.
 - **D** Carbon occurs naturally as a diatomic molecule.

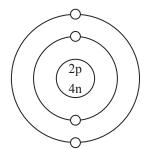
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5. Which diagram shows the structure of a $\frac{4}{2}$ He atom?

A



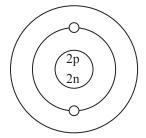
C



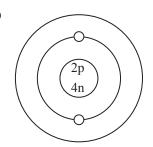
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- p protonO electron
- n neutron

B

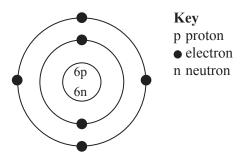


D



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6. The diagram represents an atom of an element.



Which symbol represents this atom?

- $\mathbf{A} \stackrel{9}{_4} \mathrm{Be}$
- $\mathbf{B} \quad {}^{9}_{5} \, \mathrm{B}$
- $C_{6}^{12} C$
- $\mathbf{D} = \frac{24}{12} \, \mathrm{Mg}$

6. C The atom has 6 protons (proton number = 6) and 6 neutrons (nucleon number = 6 + 6 = 12).

nucleon number of 12 and a proton number of 6, hence, it has 6 neutrons in its nucleus. 5. **B** A helium atom has 2 protons (proton number = 2) and 2 neutrons (nucleon number = 2 + 2 = 4) and 2 electrons.

4. **B** Carbon has a proton number of 6 (eliminate option C). The electronic configuration of carbon is 2.4. Hence, it has 4 electrons in its outermost shell (eliminate option A). It is naturally a giant molecule (eliminate option D). Since it has a

3. **D** The atom of element X has 8 protons, 8 electrons and 10 neutrons.

2. C An atom of element Q has 6 protons (proton number = 6) and 9 neutrons (nucleon number = 6 + 9 = 15).

I B

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Answers to